

THE STRUCTURE OF RUPESTROL - A NEW SESQUITERPENOID
FROM VERBESINA RUPESTRIS (URB.) BLAKE

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We report here the structural elucidation of a new sesquiterpene, rupestrol, which occurs naturally as its orthocinnamate (1) and cinnamate (2) esters in Verbesina rupestris.

Rupestrol orthocinnamate (1), $C_{24}H_{32}O_5$, m/e 400 (M^+), $m.p.$ 206-207°, had a trans-styrene unit [λ_{max} 250 nm (ϵ 22,000), ν_{max} 1656, 1587, 1572, 970, 750 and 700 cm^{-1} , δ (d_6 -DMSO) 6.22 and 6.90 (1H each, AB system, J_{AB} 16.5 Hz), 7.39 (5H, m, aromatic H)] and two secondary hydroxy groups [ν_{max} 3400 cm^{-1} , δ 4.67 (1H, d, J 1.5 Hz) and 4.82 (1H, d, J 2.0 Hz) vanishing on treatment with D_2O]. On mild acid treatment, it afforded rupestrol cinnamate (2) $C_{24}H_{34}O_6$, $m.p.$ 222-224°, λ_{max} 221 and 278.5 nm (ϵ 12,300 and 20,600), ν_{max} 3175, 1724, 1631 and 990 cm^{-1} , δ (d_6 -DMSO) 6.68 and 7.73 (1H each, AB system, J_{AB} 16 Hz), 7.55 (5H, m, aromatic H). Basic hydrolysis of (2) gave cinnamic acid together with the parent sesquiterpene rupestrol (3), $C_{15}H_{28}O_5$, $m.p.$ 206-207°, $[\alpha]_D + 11^\circ$, ν_{max} 3150 cm^{-1} . The NMR spectra of rupestrol and its derivatives indicated the presence of one tertiary and two secondary methyl groups.

Acetylation of (2) with acetic anhydride-pyridine gave the amorphous triacetate (4), ν_{max} 3460, 1725 cm^{-1} , δ ($CDCl_3$) 4.20 and 4.58 (1H each, d, J 12 Hz, CH_2O-), 4.08 (OH), 5.61 (1H, dd, J 11, 3 Hz, H-6) and 4.73-5.06 (2H, m, H-1 and H-9). Hydrolysis of (4) with one mole of potassium hydroxide in methanol (reflux, 2 hr) afforded the diacetate (5), $C_{19}H_{32}O_7$,

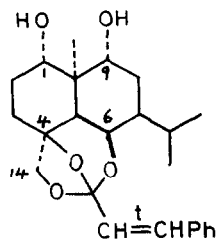
m.p. 185-186°, ν_{\max} 3350, 1724 cm^{-1} , δ 3.84 (2H, bs, CH_2OH), 4.44 (1H, m, H-6), 4.65-5.09 (2H, m, H-1 and H-9).

Rupesterol cinnamate is inert to periodic acid. However, periodate oxidation of the diacetate (5) gave the ketone (7) as a gum, ν_{\max} 3350, 1724 cm^{-1} , δ 2.70 (1H, d, J 10.5 Hz, H-5), 4.32 (1H, dd, sharpening on deuterium exchange, J 10.5, 4 Hz, H-6). The vicinal couplings of H-6 indicate a diaxial relationship with H-5 and a small dihedral angle relationship with H-7. These assignments were confirmed in the NMR spectrum of the derived acetate (8), $\text{C}_{20}\text{H}_{30}\text{O}_7$, m.p. 137-138°, ν_{\max} 1725 cm^{-1} which had δ 2.94 (1H, d, J 11.5 Hz, H-5), 5.11 (1H, dd, J 11.5, 5 Hz, H-6), 5.35 (1H, dd, J 11, 4.5 Hz, H-1) and 5.23 (1H, dd, J 7.5, 3.5 Hz, H-9).

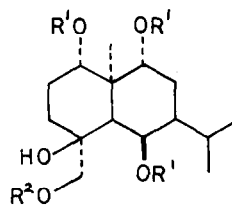
With acetone and anhydrous copper sulphate, (5) gave the acetonide (6), m.p. 152-153°, ν_{\max} 3400, 1736 cm^{-1} , in which protons on carbon bearing oxygen were clearly resolved. A singlet (2H) at δ 4.00 could be ascribed to H-14; a double doublet (J 8, 4 Hz) at 4.23 (which sharpened on deuterium exchange) to H-6. Two other double doublets at δ 4.64 (J 8, 5 Hz) and 5.05 (J 5.5, 2.5 Hz) must be attributed to H-1 and H-9. Oxidation of the acetonide (6) or the triol (5) with Jones reagent yielded the ketone (9), $\text{C}_{22}\text{H}_{34}\text{O}_7$, m.p. 184-186°, ν_{\max} 1733, 1718 cm^{-1} , δ 1.38 (6H, s, acetonide CH_3), 3.75 and 4.90 (1H each, AB system, J 7.5 Hz, $-\text{CH}_2\text{O}-$), 3.03 (1H, s, H-5), 4.90-5.45 (2H, m, H-1 and H-9).

Treatment of (9) with 60% aqueous acetic acid gave the amorphous diol (10), ν_{\max} 3300, 1735, 1695 cm^{-1} , δ 2.83 (1H, s, H-5), 3.55 and 4.30 (1H each, AB system, J 11.5 Hz, CH_2OH) which on oxidation with sodium periodate afforded the β -diketone (11), $\text{C}_{18}\text{H}_{26}\text{O}_6$, m.p. 101-102°, λ_{\max} 290.5 nm (ϵ 9,700) shifting on addition of base to 309.5 nm (ϵ 12,300), ν_{\max} 1727, 1578 cm^{-1} , δ 4.83-5.10 (2H, m, H-1 and H-9), 16.74 (OH of enolic form).

The above evidence confirms that rupesterol is a pentahydroxy sesquiterpene of the eudesmane type with hydroxy groups at C-4, C-14 and C-6 and two other secondary alcohols. The multiplicities of the protons at the bases of the acetates in (6) and (8) require that each has two adjacent protons



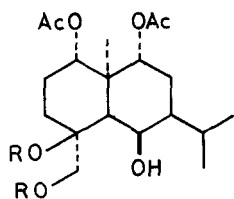
1



2 $R^1=H, R^2=\text{cinnamoyl}$

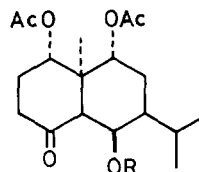
3 $R^1=R^2=H$

4 $R^1=Ac, R^2=\text{cinnamoyl}$



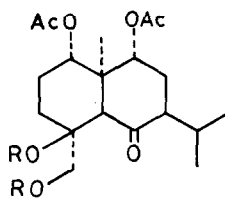
5 $R=H$

6 $R=CMe_2$



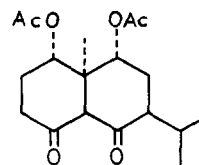
7 $R=H$

8 $R=Ac$

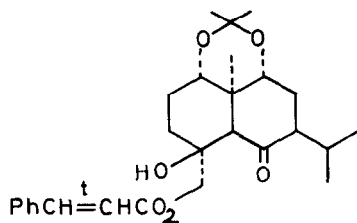


9 $R=CMe_2$

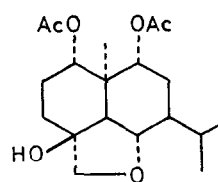
10 $R=H$



11



12



13

and confirm the location of the secondary alcohols at C-1 and C-9: C-3 is excluded by the inertness of (2) to periodic acid (see above). The location of the hydroxy groups at C-1 and C-9 was further supported by the isolation of the keto acetonide (12) $C_{27}H_{36}O_6$, from Jones oxidation of (2). This had UV and IR bands characteristic of the cinnamate moiety and the structure was confirmed by the NMR spectrum: δ 1.46 and 1.55 (3H each, s, acetonide CH_3), 2.56 (1H, s, H-5), 4.50 and 5.10 (1H each, AB system, J 12 Hz, $-CH_2O-$), 3.10 (1H, OH , disappearing on D_2O exchange).

The relative configuration at C-4 was revealed by the preparation of the ether (13), $C_{19}H_{30}O_6$, m.p. 164-165°, by the treatment of (5) with anhydrous cupric sulphate-conc. sulphuric acid. The NMR spectrum of (13) showed H-6 as a well resolved triplet (J 5.3 Hz) at δ 4.26 indicating that H-6 is now equatorial. Thus the ether linkage at C-6 and the C-4 - C-14 bond must both be axial.

The absolute configuration of rupesterol was disclosed by the CD curve of the ketone (8) ($\Delta\epsilon_{294\text{ nm}} +1.99$). Further evidence for the stereochemistry of rupesterol will be discussed in the full paper.

Rupesterol is thus revealed as a highly oxygenated enantio-eudesmane containing the rare oxygenation at C-9.¹ Eudesmane sesquiterpenes have also been isolated from Verbesina virginica.²

REFERENCES

1. F. Bohlmann and M. Grenz, Tetrahedron Letters, 5111 (1969); H. Wada, Y. Shizuri, K. Sugiura, K. Yamada and Y. Hirata, ibid., 3131 (1971).
2. P.D. Gardner, G.J. Park and C.C. Albers, J.Amer.Chem.Soc., **83**, 1511 (1961).